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ADDITION OF SULPHENYL CHLORIDES TO ACETYLENES. STRUCTURE OF THE INTERMEDIATE

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ADDITION OF SULPHENYL CHLORIDES TO ACETYLENES. STRUCTURE OF THE INTERMEDIATE.

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Sulphenyl chlorides react in sulphur dioxide with acetylenes to give thiirenium chlorides $\underline{1}$ which lead to the final adducts $\underline{2}$.

$$-C \equiv C - + R-S-C1$$

$$C = C + C1$$

$$R-S$$

$$1$$

$$R$$

$$\frac{1}{R}$$

$$\frac{2}{R}$$

Thiirenium ions have been detected in this solvent at low temperature and in some cases their hexachloroantimonate or tetrafluoroborate salts isolated. Evidence for the intervention of a different intermediate, probably the sulphurane 3, in reactions carried out in differnt, less polar solvents will be discussed.

$$C \longrightarrow C$$
 R
 $C1$

The effect of the solvent on the orientation of the addition of sulphenyl chlorides to asymmetric acetylenes will be also brefly rehexamined.